

Chapter Test B Arrangement Of Electrons In Atoms

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Modern Chemistry 31 Chapter Test Chapter: Arrangement of Electrons in Atoms PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. The principal quantum number of an electron is 4. What are the

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Modern Chemistry 28 Chapter Test Name Class Date Chapter Test A, continued ____ 8. What is the frequency of a photon whose energy is $3.4 \times 10^{19} \text{ J}$? (h $6.626 \times 10^{-34} \text{ J}\cdot\text{s}$) a. $8.8 \times 10^{26} \text{ Hz}$ b. $5.1 \times 10^{14} \text{ Hz}$ c. $1.9 \times 10^{-15} \text{ Hz}$ d. $2.3 \times 10^{-52} \text{ Hz}$ ____ 9. When electromagnetic radiation strikes the surface of a metal, electrons are ejected from the metal's ...

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(b) particle nature. (d) wave nature. 6. c According to Bohr, which is the point in the figure below where electrons cannot reside? (a) point A (c) point C (b)

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point B (d) point D 7. c According to the quantum theory, point D in the above figure represents (a) the fixed position of an electron. (b) the farthest position from the nucleus that an ...

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_____ refers to the arrangement of conditions in a study so that meaningful comparisons of the effect of the IV can be made 1. intervention 2. experimental design 3. single-subject design 4. group comparison design

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